

Periodic Table of the Elements

1 1.0080 ±1 20.28 17.81 0.08988 1s ¹ hydrogen																	18 4.002602 0 4.216 0.8 (26 bar) 0.1787 1s ² helium				
3 6.97 1615 453.69 0.53 [He]2s ¹ lithium		4 9.012182 2745 1560 1.848 [He]2s ² beryllium																		10 20.1797 27.102 53.53 24.48 0.8999 [He]2s ² 2p ⁶ neon	
11 22.989769 1156.1 370.96 0.971 [Ne]3s ¹ sodium		12 24.3050 1363 922 1.738 [Ne]3s ² magnesium																		18 39.948 87.5 172.17 84 1.7824 [Ne]3s ² 3p ⁶ argon	
19 39.0983 1033.1 336.4 0.862 [Ar]4s ¹ potassium	20 40.078 1757 1112 1.55 [Ar]4s ² calcium	21 44.955912 3103 1814 3.00 [Ar]4s ² 3d ¹ scandium	22 47.867 3560 1941±10 4.50 [Ar]4s ² 3d ² titanium	23 50.9415 3680 2163±10 5.80 [Ar]4s ² 3d ³ vanadium	24 51.9961 2945 2130 7.19 [Ar]4s ¹ 3d ⁵ chromium	25 54.938045 2334 1517 7.43 [Ar]4s ² 3d ⁵ manganese	26 55.845 3134 1808 7.86 [Ar]4s ² 3d ⁶ iron	27 58.933195 3200 1768 8.90 [Ar]4s ² 3d ⁷ cobalt	28 58.6934 3186 1726 8.90 [Ar]4s ² 3d ⁸ nickel	29 63.546 2840 1356 8.96 [Ar]4s ¹ 3d ¹⁰ copper	30 65.38 1180 692.73 7.14 [Ar]4s ² 3d ¹⁰ zinc	31 69.723 2477 302.93 5.907 [Ar]4s ² 3d ¹⁰ 4p ¹ gallium	32 72.63 2353 302.93 5.323 [Ar]4s ² 3d ¹⁰ 4p ² germanium	33 74.92160 886 (sub.) 1000 (28 bar) 5.72 [Ar]4s ² 3d ¹⁰ 4p ³ arsenic	34 78.96 958.1 490 4.79 [Ar]4s ² 3d ¹⁰ 4p ⁴ selenium	35 79.904 331.93 266 3.119 [Ar]4s ² 3d ¹⁰ 4p ⁵ bromine	36 83.798 120.9 116.6 3.708 [Ar]4s ² 3d ¹⁰ 4p ⁶ krypton				
37 85.4678 959 312.04 1.53 [Kr]5s ¹ rubidium	38 87.62 1657 1042 2.60 [Kr]5s ² strontium	39 88.90585 3611 1795±8 4.47 [Kr]5s ² 4d ¹ yttrium	40 91.224 4650 2125±2 6.40 [Kr]5s ² 4d ² zirconium	41 92.906 5015 2471±10 8.57 [Kr]5s ¹ 4d ⁴ niobium	42 95.96 4885 2890 10.20 [Kr]5s ¹ 4d ⁵ molybdenum	43 [97.9072] 5150 2445 11.5 [Kr]5s ² 4d ⁵ technetium	44 101.07 4423 2607 12.20 [Kr]5s ¹ 4d ⁷ ruthenium	45 102.90550 3968 2239±3 12.40 [Kr]5s ¹ 4d ⁸ rhodium	46 106.42 3213 1825 12.02 [Kr]4d ¹⁰ palladium	47 107.8682 2435 1235.08 10.50 [Kr]5s ¹ 4d ¹⁰ silver	48 112.411 1038 594.1 8.65 [Kr]5s ² 4d ¹⁰ cadmium	49 114.818 2353 429.76 7.31 [Kr]5s ² 4d ¹⁰ 5p ¹ indium	50 118.710 2875 505.12 7.30 [Kr]5s ² 4d ¹⁰ 5p ² tin	51 121.760 1860 903.89 6.684 [Kr]5s ² 4d ¹⁰ 5p ³ antimony	52 127.60 1263.1 722.7 6.24 [Kr]5s ² 4d ¹⁰ 5p ⁴ tellurium	53 126.90447 487.35 (35 bar) 387 4.93 [Kr]5s ² 4d ¹⁰ 5p ⁵ iodine	54 131.293 166.1 161.3 5.88 [Kr]5s ² 4d ¹⁰ 5p ⁶ xenon				
55 132.905452 942.5 301.55 1.873 [Xe]6s ¹ cesium	56 137.327 2170 998 3.51 [Xe]6s ² barium	57 174.9668 3675 1936 9.85 [Xe]6s ² 4f ¹⁴ 5d ¹ lutetium	72 178.49 4875 2500 13.2 [Xe]6s ² 4f ¹⁴ 5d ² hafnium	73 180.94788 5700±100 3269 16.6 [Xe]6s ² 4f ¹⁴ 5d ³ tantalum	74 183.84 5933 3683±20 19.3 [Xe]6s ² 4f ¹⁴ 5d ⁴ tungsten	75 186.207 5900 (est.) 3453 21.0 [Xe]6s ² 4f ¹⁴ 5d ⁵ rhenium	76 190.23 5300 3327 22.4 [Xe]6s ² 4f ¹⁴ 5d ⁶ osmium	77 192.217 4403 2683 22.42 [Xe]6s ² 4f ¹⁴ 5d ⁷ iridium	78 195.084 4100 2045 21.45 [Xe]6s ² 4f ¹⁴ 5d ⁸ platinum	79 196.966569 3081 1937.58 13.546 [Xe]6s ² 4f ¹⁴ 5d ¹⁰ gold	80 200.59 629.73 234.28 13.546 [Xe]6s ² 4f ¹⁴ 5d ¹⁰ mercury	81 204.384 1730±10 576.7 11.85 [Xe]6s ² 4f ¹⁴ 5d ¹⁰ 6p ¹ thallium	82 207.2 2013 600.652 11.34 [Xe]6s ² 4f ¹⁴ 5d ¹⁰ 6p ² lead	83 208.98040 1833±5 544.5 9.80 [Xe]6s ² 4f ¹⁴ 5d ¹⁰ 6p ³ bismuth	84 [208.9824] 1235 527 9.4 [Xe]6s ² 4f ¹⁴ 5d ¹⁰ 6p ⁴ polonium	85 [209.9871] 610 575 — [Xe]6s ² 4f ¹⁴ 5d ¹⁰ 6p ⁵ astatine	86 [222.0176] 211 202 9.73 [Xe]6s ² 4f ¹⁴ 5d ¹⁰ 6p ⁶ radon				
87 [223.0197] 950 300 [Rn]7s ¹ francium	88 [226.0254] 1413 973 5.0 [Rn]7s ² radium	103 [261.1096] — — [Rn]7s ² 5f ¹⁴ 6d ¹ lawrencium	104 [265.1167] — — [Rn]7s ² 5f ¹⁴ 6d ² rutherfordium	105 [268.1250] — — [Rn]7s ² 5f ¹⁴ 6d ³ dubnium	106 [271.133] — — [Rn]7s ² 5f ¹⁴ 6d ⁴ seaborgium	107 [270] — — [Rn]7s ² 5f ¹⁴ 6d ⁵ bohrium	108 [277.150] — — [Rn]7s ² 5f ¹⁴ 6d ⁶ hassium	109 [276.151] — — [Rn]7s ² 5f ¹⁴ 6d ⁷ meitnerium	110 [281] — — [Rn]7s ² 5f ¹⁴ 6d ⁸ darmstadtium	111 [280.164] — — [Rn]7s ² 5f ¹⁴ 6d ⁹ roentgenium	112 [285.174] — — [Rn]7s ² 5f ¹⁴ 6d ¹⁰ copernicium	113 [285] provisional [Rn]7s ² 5f ¹⁴ 6d ¹⁰ 7p ¹ nihonium	114 [289.187] [289] provisional [Rn]7s ² 5f ¹⁴ 6d ¹⁰ 7p ² flerovium	115 [288] provisional [Rn]7s ² 5f ¹⁴ 6d ¹⁰ 7p ³ moscovium	116 [293] [294] provisional [Rn]7s ² 5f ¹⁴ 6d ¹⁰ 7p ⁴ livermorium	117 [294] provisional [Rn]7s ² 5f ¹⁴ 6d ¹⁰ 7p ⁵ tennessine	118 [294] provisional [Rn]7s ² 5f ¹⁴ 6d ¹⁰ 7p ⁶ oganeson				

atomic number → 26
atomic mass → 55.845
boiling point /K → 3134
melting point /K → 1808
density /g/cm³, g/L for gases → 7.86
name → iron
symbol → Fe
common oxidation states (most stable) → 2, (3)
electronic configuration → [Ar]4s²3d⁶
solid, liquid, gas, synthetic

All properties at 298.15 K and 1 bar unless noted.

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Data from www.iupac.org. For several elements, the variability in atomic mass depends substantially on whether the sample origin is organic or inorganic. For these elements, the average, with decreased precision, is reported.
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57 138.90547 3737 1191 6.7 [Xe]6s ² 5d ¹ lanthanum	58 140.116 (3),4 3706 1071±3 6.78 [Xe]6s ² 4f ¹ 5d ¹ cerium	59 140.90765 (3),4 3793 1204 6.77 [Xe]6s ² 4f ³ praseodymium	60 144.242 3347 1294 7.00 [Xe]6s ² 4f ⁴ neodymium	61 [144.9127] 3300 (est.) 1315 6.475 [Xe]6s ² 4f ⁶ promethium	62 150.36 2067 1347 7.54 [Xe]6s ² 4f ⁶ samarium	63 151.964 1800 1095 5.259 [Xe]6s ² 4f ⁷ europium	64 157.25 3546 1586 7.895 [Xe]6s ² 4f ⁷ 5d ¹ gadolinium	65 158.925 3503 1629 8.27 [Xe]6s ² 4f ⁹ terbium	66 162.500 2840 1685 8.536 [Xe]6s ² 4f ¹⁰ dysprosium	67 164.93032 2973 1747 8.80 [Xe]6s ² 4f ¹¹ holmium	68 167.259 3141 1802 9.05 [Xe]6s ² 4f ¹² erbium	69 168.93421 2223 1818 9.33 [Xe]6s ² 4f ¹³ thulium	70 173.054 1469 1092 6.98 [Xe]6s ² 4f ¹⁴ ytterbium
89 [227.0278] 3500±300 1323 10.07 [Rn]7s ² 6d ¹ actinium	90 232.0381 4273 2023 11.70 [Rn]7s ² 6d ² thorium	91 [231.0359] — 1873 15.40 [Rn]7s ² 5f ⁶ 6d ¹ protactinium	92 238.02891 4091 1405 18.90 [Rn]7s ² 5f ⁶ 6d ¹ uranium	93 [237.0482] 4175 913 20.45 [Rn]7s ² 5f ⁶ 6d ¹ neptunium	94 [244.0642] 3505 914 19.80 [Rn]7s ² 5f ⁶ plutonium	95 [243.0614] 2880 1267 13.6 [Rn]7s ² 5f ⁷ americium	96 [247.0704] 1613 13.5 [Rn]7s ² 5f ⁷ 6d ¹ curium	97 [247.0703] — — [Rn]7s ² 5f ⁹ berkelium	98 [251.0796] — — [Rn]7s ² 5f ¹⁰ californium	99 [252.0830] — — [Rn]7s ² 5f ¹¹ einsteinium	100 [257.0951] — — [Rn]7s ² 5f ¹² fermium	101 [258.0984] — — [Rn]7s ² 5f ¹³ mendelevium	102 [259.1010] — — [Rn]7s ² 5f ¹⁴ nobelium

Periodic Table of the Elements

1																												18											
1 H																		2 He																					
25 154 (-1) — 72.77 13.598 — 2.20																		30 0.0845 — — 20.79 0.152 — 24.587 54.418 0																					
3 Li		4 Be																5 B		6 C		7 N		8 O		9 F		10 Ne											
145 78 (+1) — 59.63 5.3915 75.639		105 34 (+2) — 16.40 — 200 31.30 1.57																85 23 (+3) — 26.99 8.2976 25.155		70 16 (+4) — 8.53 — 2.04		65 171 (-3) — 29.13 — 3.04		60 22 (+1) — 29.44 — 3.44		50 133 (-1) — 31.16 — 0.0279 3.98		40 1.7326 — 0.3317 — 18.24 — 0.0493 — 21.565 — 40.963											
11 Na		12 Mg																13 Al		14 Si		15 P		16 S		17 Cl		18 Ar											
180 98 (+1) — 52.87 5.1386 47.282		150 79 (+2) — 156 22.60 1.31																125 57 (+3) — 41.76 5.9854 18.829		110 26 (+4) — 134.07 8.1515 16.345		100 17 (+5) — 212 (-3) — 2.19		100 29 (+6) — 22.77 — 3.16		100 181 (-1) — 34.03 — 0.1772 — 15.760 — 27.629													
19 K		20 Ca		21 Sc		22 Ti		23 V		24 Cr		25 Mn		26 Fe		27 Co		28 Ni		29 Cu		30 Zn		31 Ga		32 Ge		33 As		34 Se		35 Br		36 Kr					
220 133 (+1) — 48.39 4.3406 31.632		180 106 (+2) — 200 29.80 1.00		160 83 (+3) — 26.97 — 1.36		140 86 (+2) — 21.9 — 1.54		135 79 (+2) — 50.66 6.7461 14.655		140 82 (+2) — 64.26 6.7668 16.485		140 82 (+2) — 93.7 — 1.66		140 91 (+2) — 7.82 — 1.55		140 82 (+2) — 14.57 7.9028 16.188		135 82 (+2) — 63.87 — 1.88		135 78 (+2) — 111.54 7.6395 18.169		135 96 (+1) — 119.16 59.60 1.90		135 83 (+2) — 116 16.60 1.65		130 113 (+2) — 40.6 6.78 1.81		125 53 (+4) — 59.9 — 2.18		115 69 (+3) — 78.54 9.7528 2.195		115 69 (+4) — 2.04 — 2.55		115 196 (-1) — 0.122 — 2.96		90 9.029 — 1.638 — 0.00949 — 14.000 — 24.360			
37 Rb		38 Sr		39 Y		40 Zr		41 Nb		42 Mo		43 Tc		44 Ru		45 Rh		46 Pd		47 Ag		48 Cd		49 In		50 Sn		51 Sb		52 Te		53 I		54 Xe					
235 149 (+1) — 46.88 4.1768 27.289		200 127 (+2) — 35.3 7.62 0.95		180 106 (+3) — 26.67 — 1.22		155 109 (+2) — 24.63 — 1.33		145 74 (+4) — 26.4 — 1.60		145 69 (+5) — 24.16 — 1.70		145 62 (+6) — 23.99 — 1.86		135 72 (+5) — 20.56 — 1.90		130 65 (+4) — 24.05 — 2.20		135 67 (+4) — 24.90 — 2.28		140 89 (+5) — 25.54 — 2.28		160 89 (+2) — 25.35 — 1.93		155 103 (+2) — 25.85 — 1.69		155 132 (+1) — 26.41 — 1.78		145 93 (+2) — 26.95 — 2.05		145 74 (+4) — 26.95 — 2.05		140 62 (+5) — 25.57 — 2.10		140 97 (+4) — 17.49 — 2.66		140 196 (-1) — 7.824 — 2.297		110 12.636 — 2.297 — 0.00569 — 12.130 — 21.209	
55 Cs		56 Ba		71 Lu		72 Hf		73 Ta		74 W		75 Re		76 Os		77 Ir		78 Pt		79 Au		80 Hg		81 Tl		82 Pb		83 Bi		84 Po		85 At		86 Rn					
260 165 (+1) — 45.51 3.8939 23.157		215 143 (+2) — 28.01 — 0.89		175 85 (+3) — 26.25 — 1.27		155 84 (+4) — 24.06 — 1.30		145 72 (+3) — 31.6 — 1.50		135 64 (+5) — 25.33 — 1.50		135 62 (+6) — 23.90 — 2.36		135 60 (+7) — 33.2 — 1.90		130 67 (+4) — 31.8 — 2.20		135 75 (+3) — 26.1 — 2.28		135 70 (+4) — 19.6 — 2.28		135 137 (+1) — 12.55 — 2.54		150 112 (+2) — 2.295 — 2.00		190 149 (+1) — 4.142 — 2.04		180 132 (+2) — 4.799 — 1.87		160 96 (+3) — 11.3 — 2.02		190 65 (+4) — 25.08 — 2.00		190 230 (-2) — 25.08 — 2.00		190 57 (+5) — 2.89 — 0.00364 — 16.4 — 2.89 — 19.98 — 16.4 — 2.89 — 0			
87 Fr		88 Ra		103 Lr		104 Rf		105 Db		106 Sg		107 Bh		108 Hs		109 Mt		110 Ds		111 Rg		112 Cn		113 Nh		114 Fl		115 Mc		116 Lv		117 Ts		118 Og					
180 (+1) — — 44.38 3.9384 — 0.70		215 162 (+2) — 27.12 — 0.90		88 (+3) — — 10 — —		— — — 23 — —		— — — 58 — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —		— — — — — —					

atomic radius /pm → 140 349.6
 ionic radius (charge) /pm → 82 (+2) 13.8
 ionic radius (charge) /pm → 67 (+3) 24.57
 electron affinity / (kJ/mol) → 14.57 80.2
 first ionization energy /eV → 7.9028 9.93
 second ionization energy /eV → 16.188 1.83

enthalpy of vaporization / (kJ/mol) ← 349.6
 enthalpy of fusion / (kJ/mol) ← 13.8
 molar heat capacity / (J/(mol·K)) ← 24.57
 thermal conductivity / (W/(m·K)) ← 80.2
 electrical conductivity / (10⁹/(m·Ω)) ← 9.93
 electronegativity (Pauling) ← 1.83

All properties at 298.15 K and 1 bar unless noted.

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Data from www.iupac.org. For several elements, the variability in atomic mass depends substantially on whether the sample origin is organic or inorganic. For these elements, the average, with decreased precision, is reported.

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* 57 La		58 Ce		59 Pr		60 Nd		61 Pm		62 Sm		63 Eu		64 Gd		65 Tb		66 Dy		67 Ho		68 Er		69 Tm		70 Yb					
195 122 (+3) — 45.35 5.5770 11.059		185 107 (+3) — 11.4 5.5387 10.882		185 106 (+3) — 12.5 5.4620 10.572		185 104 (+3) — 13.3 5.2522 10.779		185 109 (+3) — 17.9 5.5967 10.882		185 108 (+3) — 13.9 5.6433 11.090		185 89 (+3) — 23.35 5.6703 11.245		180 97 (+3) — 10.6 6.1502 12.126		175 81 (+4) — 11.1 5.8641 11.504		175 81 (+4) — 10.7 5.9387 11.712		175 89 (+3) — 10.7 6.0216 11.815		175 89 (+3) — 12.2 6.1077 12.223		175 89 (+3) — 19.9 6.1844 12.023		175 94 (+3) — 16.84 6.2538 12.176		175 87 (+4) — 27.03 — 12.176		175 86 (+3) — 25.96 — —	
** 89 Ac		90 Th		91 Pa		92 U		93 Np		94 Pu		95 Am		96 Cm		97 Bk		98 Cf		99 Es		100 Fm		101 Md		102 No					
195 162 (+2) — 33.77 5.1718 12.126		180 99 (+3) — 27.84 6.0838 11.504		180 113 (+3) — 12.3 5.8869 1.50		175 103 (+3) — 8.52 6.1937 14.717		175 110 (+3) — 5.19 28.45 —		175 108 (+3) — 2.84 31.73 —		175 82 (+6) — 26.74 — —		175 81 (+6) — 6.74 — —		175 107 (+3) — 14.4 — —		175 80 (+6) — 26.74 — —		175 99 (+3) — 15 — —		175 98 (+3) — — — —		175 98 (+3) — — — —		175 91 (+3) — — — —		175 90 (+3) — — — —		175 112 (+2) — — — —	